Chemical Safety Data Sheet MSDS / SDS

POE (2) TALLOW AMINE

Revision Date:2025-07-05 Revision Number:1

SECTION 1: Identification of the substance/mixture and of the company/undertaking

Product identifier

Product name	: POE (2) TALLOW AMINE	
CBnumber	: CB2506339	
CAS	: 61791-44-4	
EINECS Number	: 263-177-5	
Synonyms	: Bis (2-hydroxyethyl) tallow alkylamine,EthomeenT/12	
Relevant identified uses of the substance or mixture and uses advised against		
Relevant identified uses	: For R&D use only. Not for medicinal, household or other use.	
Uses advised against	: none	
Company Identification		
Company	: Chemicalbook	
Address	: Building 1, Huihuang International, Shangdi 10th Street, Haidian District, Beijing	
Telephone	: 010-86108875	

SECTION 2: Hazards identification

Classification of the substance or mixture

Acute toxicity - Category 4, Oral Skin corrosion, Sub-category 1C Serious eye damage, Category 1 Hazardous to the aquatic environment, short-term (Acute) - Category Acute 1 Hazardous to the aquatic environment, long-term (Chronic) - Category Chronic 2

Label elements

Pictogram(s)

Signal word

Danger

Hazard statement(s)

H302 Harmful if swallowed

H314 Causes severe skin burns and eye damage

H318 Causes serious eye damage

H400 Very toxic to aquatic life

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H411 Toxic to aquatic life with long lasting effects

Precautionary statement(s)

Prevention

P264 Wash ... thoroughly after handling.

P270 Do not eat, drink or smoke when using this product.

P260 Do not breathe dust/fume/gas/mist/vapours/spray.

P280 Wear protective gloves/protective clothing/eye protection/face protection/hearing protection/...

P273 Avoid release to the environment.

Response

P301+P317 IF SWALLOWED: Get medical help.

P330 Rinse mouth.

P301+P330+P331 IF SWALLOWED: Rinse mouth. Do NOT induce vomiting.

P363 Wash contaminated clothing before reuse.

P304+P340 IF INHALED: Remove person to fresh air and keep comfortable for breathing.

P316 Get emergency medical help immediately.

P321 Specific treatment (see ... on this label).

P305+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue

rinsing.

P305+P354+P338 IF IN EYES: Immediately rinse with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

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P317 Get medical help.

P391 Collect spillage.

Storage

P405 Store locked up.

Disposal

P501 Dispose of contents/container to an appropriate treatment and disposal facility in accordance with applicable laws and regulations, and product characteristics at time of disposal.

Other hazards

no data available

SECTION 3: Composition/information on ingredients

Substance

Product name	: POE (2) TALLOW AMINE
Synonyms	: Bis (2-hydroxyethyl) tallow alkylamine, EthomeenT/12
CAS	: 61791-44-4
EC number	: 263-177-5
MF	: C4H11NO2
MW	: 105.136

Description of first aid measures

If inhaled

Fresh air, rest.

Following skin contact

Remove contaminated clothes. Rinse skin with plenty of water or shower.

Following eye contact

First rinse with plenty of water for several minutes (remove contact lenses if easily possible), then refer for medical attention.

Following ingestion

Rinse mouth. Give one or two glasses of water to drink. Refer for medical attention . Rest.

Most important symptoms and effects, both acute and delayed

Irritation of eyes and skin. Breathing vapors may cause coughing, a smothering sensation, nausea, headache. (USCG, 1999)

Indication of any immediate medical attention and special treatment needed

Immediate first aid: Ensure that adequate decontamination has been carried out. If patient is not breathing, start artificial respiration, preferably with a demand valve resuscitator, bag-valve-mask device, or pocket mask, as trained. Perform CPR if necessary. Immediately flush contaminated eyes with gently flowing water. Do not induce vomiting. If vomiting occurs, lean patient forward or place on the left side (head-down position, if possible) to maintain an open airway and prevent aspiration. Keep patient quiet and maintain normal body temperature. Obtain medical attention. /Organic bases/Amines and related compounds/

SECTION 5: Firefighting measures

Extinguishing media

Wear self contained breathing apparatus for fire fighting if necessary.

Specific Hazards Arising from the Chemical

Special Hazards of Combustion Products: Irritating vapors are generated when heated. (USCG, 1999)

Advice for firefighters

Wear self-contained breathing apparatus for firefighting if necessary.

SECTION 6: Accidental release measures

Personal precautions, protective equipment and emergency procedures

Avoid dust formation. Avoid breathing mist, gas or vapours. Avoid contacting with skin and eye. Use personal protective equipment. Wear chemical impermeable gloves. Ensure adequate ventilation. Remove all sources of ignition. Evacuate personnel to safe areas. Keep people away from and upwind of spill/leak.

Environmental precautions

Personal protection: filter respirator for organic gases and particulates adapted to the airborne concentration of the substance. Sweep spilled substance into covered sealable containers. If appropriate, moisten first to prevent dusting. Then store and dispose of according to local

Methods and materials for containment and cleaning up

Personal precautions, protective equipment and emergency procedures: Use personal protective equipment. Avoid breathing vapours, mist or gas. Ensure adequate ventilation. Evacuate personnel to safe areas.

SECTION 7: Handling and storage

Precautions for safe handling

Handling in a well ventilated place. Wear suitable protective clothing. Avoid contact with skin and eyes. Avoid formation of dust and aerosols. Use non-sparking tools. Prevent fire caused by electrostatic discharge steam.

Conditions for safe storage, including any incompatibilities

Separated from strong oxidants and acids. Dry.Keep container tightly closed in a dry and well-ventilated place. Containers which are opened must be carefully resealed and kept upright to prevent leakage. Air sensitive.

SECTION 8: Exposure controls/personal protection

Control parameters

Occupational Exposure limit values

Component	Ethanol, 2,2'-iminobis-, N-tallow alkyl derivs.	
CAS No.	61791-44-4	
	Recommended Exposure Limit: 10 Hour Time-Weighted Average: 3 ppm (15 mg/cu m).	

Biological limit values

no data available

Exposure controls

Ensure adequate ventilation. Handle in accordance with good industrial hygiene and safety practice. Set up emergency exits and the riskelimination area.

Individual protection measures

Eye/face protection

Wear tightly fitting safety goggles with side-shields conforming to EN 166(EU) or NIOSH (US).

Skin protection

Wear fire/flame resistant and impervious clothing. Handle with gloves. Gloves must be inspected prior to use. Wash and dry hands. The

selected protective gloves have to satisfy the specifications of EU Directive 89/686/EEC and the standard EN 374 derived from it.

Respiratory protection

If the exposure limits are exceeded, irritation or other symptoms are experienced, use a full-face respirator.

Thermal hazards

no data available

Information on basic physicochemical properties

Physical state	Diethanolamine is an oily colorless liquid or solid white crystals. Slight rotten fish or ammonia odor.
	Denser than water. (USCG, 1999)
Colour	A faintly colored, viscous liquid or deliquescent prisms
Odour	Mild, ammonia-like odor
Melting point/freezing point	28°C
Boiling point or initial boiling point and	239.4°C at 760 mmHg
boiling range	
Flammability	Class IIIB Combustible Liquid: FI.P. at or above 200°F.Combustible Solid
Lower and upper explosion	Lower flammable limit: 1.6% by volume (calculated); Upper flammable limit: 9.8% by volume
limit/flammability limit	(estimated)
Flash point	279° F (NTP, 1992)
Auto-ignition temperature	1224° F (USCG, 1999)
Decomposition temperature	no data available
рН	Strong base. pH of 0.1 N aqueous solution: 11.0
Kinematic viscosity	351.9 cP at 30 deg C; 53.85 cP at 60 deg C
Solubility	greater than or equal to 100 mg/mL at 57° F (NTP, 1992)
Partition coefficient n-octanol/water	log Kow= - 1.43
Vapour pressure	5 mm Hg at 280° F ; <0.01 mm Hg at 68° F (NTP, 1992)
Density and/or relative density	0.996
Relative vapour density	3.65 (NTP, 1992) (Relative to Air)
Particle characteristics	no data available

SECTION 10: Stability and reactivity

Reactivity

Decomposes on burning. This produces toxic fumes. The solution in water is a medium strong base. Reacts violently with strong oxidants and strong acids. Attacks copper, zinc, aluminium and their alloys.

Chemical stability

no data available

Possibility of hazardous reactions

Combustile, when exposed to heat or flame; can react with oxidizing materials. The vapour is heavier than air.DIETHANOLAMINE is an aminoalcohol. Amines are chemical bases. They neutralize acids to form salts plus water. These acid-base reactions are exothermic. The amount of heat that is evolved per mole of amine in a neutralization is largely independent of the strength of the amine as a base. Amines may be incompatible with isocyanates, halogenated organics, peroxides, phenols (acidic), epoxides, anhydrides, and acid halides. Flammable gaseous hydrogen is generated by amines in combination with strong reducing agents, such as hydrides. This compound is hygroscopic. It may be sensitive to exposure to air and light. This compound can react with oxidizing materials, acids, CO2, copper alloys, aluminum, zinc, galvanized iron and copper. (NTP, 1992)

Conditions to avoid

no data available

Incompatible materials

DEA (Diethanolamine) degrades in the presense of carbon dioxide to yield HEOD [3-(2-hydroxy-ethyl)oxazolidone], THEED [N,N,N,'-(2-hydroxyethyl)ethyleen-diamine] and BHEP [N,N'-bis(2-hydroxyethyl)piperazine.

Hazardous decomposition products

Special hazards arising from the substance or mixture: Carbon oxides, nitrogen oxides (NOx). Nature of decomposition products not known.

SECTION 11: Toxicological information

Acute toxicity

- Oral: LD50 Rat oral 710 mg/kg
- Inhalation: no data available
- Dermal: no data available

Skin corrosion/irritation

no data available

Serious eye damage/irritation

no data available

Respiratory or skin sensitization

no data available

Germ cell mutagenicity

no data available

Carcinogenicity

Evaluation: Cancer in humans: There is inadequate evidence in humans for the carcinogenicity of diethanolamine. Cancer in experimental animals: There is sufficient evidence in experimental animals for the carcinogenicity of diethanolamine. Overall evaluation: Diethanolamine is possibly carcinogen to humans (Group 2B).

Reproductive toxicity

No information is available on the reproductive or developmental effects of diethanolamine in humans. Animal studies have reported testicular degeneration and reduced sperm motility and count from oral exposure to diethanolamine. (10)

STOT-single exposure

no data available

STOT-repeated exposure

no data available

Aspiration hazard

SECTION 12: Ecological information

Toxicity

Toxicity to fish: LC50; Species: Lepomis macrochirus (Bluegill) weight 5 g, length 7 (5-11) cm; Conditions: freshwater, static, 20 deg C, pH 6.9-7.5, hardness 84.0-163 mg/L CaCO3, alkalinity 33.0-81.0 mg/L CaCO3, dissolved oxygen >5 mg/L; Concentration: 2100000 ug/L for 24 hr /formulation

Toxicity to daphnia and other aquatic invertebrates: LC50; Species: Daphnia magna (Water Flea) age < or =24 hr; Conditions: freshwater, static, 20-22 deg C, pH 7.6-7.7; Concentration: 180000 ug/L for 24 hr /formulation

Toxicity to algae: no data available

Toxicity to microorganisms: no data available

Persistence and degradability

AEROBIC: Biodegradation of diethanolamine has been reported in many die-away tests(1-3). N-Nitrosodiethanolamine has been identified as a metabolite of diethanolamine in natural water samples and sewage(2).Media Initial Concn Incubation Time (days) % Biodegradation Reference River Water 21 mg/L 4 5 1 River Water 210 ug/L 4 55 1 River Water 21 ng/L 4 32 1 Lake Water 1.1 ppm 14 31 2 Acidic Lake Water 1.1 ppm 14 1.2 2 Sewage 1.1 ppm 14 53 2 River Water 50 ppm 10 90 3

Bioaccumulative potential

An estimated BCF of 3 was calculated in fish for diethanolamine(SRC), using a log Kow of -1.43(1) and a regression-derived equation(2). According to a classification scheme(3), this BCF suggests the potential for bioconcentration in aquatic organisms is low(SRC).

Mobility in soil

The Koc of diethanolamine has been reported as 3.97(1) and an experimental log Koc of 0.60(2). According to a classification scheme(3), these Koc values suggest that diethanolamine is expected to have very high mobility in soil. The pKa of diethanolamine is 8.96(4), indicating that this compound will exist partially in cation form in the environment and cations generally adsorb more strongly to soils containing organic carbon and clay than their neutral counterparts(5). Diethanolamine has been shown to adsorb to humic acid which may be contained in soils and sediments(6). The adsorption of diethanolamine on humic acid changed very slightly from pH 4-8, (40-45% adsorption)(6).

Other adverse effects

no data available

SECTION 13: Disposal considerations

Disposal methods

Product

The material can be disposed of by removal to a licensed chemical destruction plant or by controlled incineration with flue gas scrubbing. Do not contaminate water, foodstuffs, feed or seed by storage or disposal. Do not discharge to sever systems.

Contaminated packaging

Containers can be triply rinsed (or equivalent) and offered for recycling or reconditioning. Alternatively, the packaging can be punctured to make it unusable for other purposes and then be disposed of in a sanitary landfill. Controlled incineration with flue gas scrubbing is possible

SECTION 14: Transport information

UN Number

ADR/RID: no data available IMDG: no data available IATA: no data available

UN Proper Shipping Name

ADR/RID: no data available IMDG: no data available IATA: no data available

Transport hazard class(es)

ADR/RID: no data available IMDG: no data available IATA: no data available

Packing group, if applicable

ADR/RID: no data available IMDG: no data available IATA: no data available

Environmental hazards

ADR/RID: Yes

IMDG: Yes

IATA: Yes

Special precautions for user

no data available

Transport in bulk according to IMO instruments

no data available

SECTION 15: Regulatory information

Safety, health and environmental regulations specific for the product in question

European Inventory of Existing Commercial Chemical Substances (EINECS)

Listed.

EC Inventory

Listed.

United States Toxic Substances Control Act (TSCA) Inventory

Listed.

China Catalog of Hazardous chemicals 2015 Not Listed. New Zealand Inventory of Chemicals (NZIoC) Listed. PICCS Listed. Vietnam National Chemical Inventory Listed. IECSC Listed. Korea Existing Chemicals List (KECL)

SECTION 16: Other information

Abbreviations and acronyms

CAS: Chemical Abstracts Service

ADR: European Agreement concerning the International Carriage of Dangerous Goods by Road

RID: Regulation concerning the International Carriage of Dangerous Goods by Rail

IMDG: International Maritime Dangerous Goods

IATA: International Air Transportation Association

TWA: Time Weighted Average

STEL: Short term exposure limit

LC50: Lethal Concentration 50%

LD50: Lethal Dose 50%

EC50: Effective Concentration 50%

References

IPCS - The International Chemical Safety Cards (ICSC), website: http://www.ilo.org/dyn/icsc/showcard.home

HSDB - Hazardous Substances Data Bank, website: https://toxnet.nlm.nih.gov/newtoxnet/hsdb.htm

IARC - International Agency for Research on Cancer, website: http://www.iarc.fr/

eChemPortal - The Global Portal to Information on Chemical Substances by OECD, website: http://www.echemportal.org/echemportal/index?

pageID=0&request_locale=en

CAMEO Chemicals, website: http://cameochemicals.noaa.gov/search/simple

ChemlDplus, website: http://chem.sis.nlm.nih.gov/chemidplus/chemidlite.jsp

ERG - Emergency Response Guidebook by U.S. Department of Transportation, website: http://www.phmsa.dot.gov/hazmat/library/erg

Germany GESTIS-database on hazard substance, website: http://www.dguv.de/ifa/gestis/gestis-stoffdatenbank/index-2.jsp

ECHA - European Chemicals Agency, website: https://echa.europa.eu/

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